

Short Communication

**Acetylene Chemistry, Part XXII [1, 2]:
Synthesis of Noracronycine and Some of its Analogs
via Mitsunobu-Reaction**

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Summary. The natural product noracronycine as well as some of its analogs were synthesized by Mitsunobu-etherisation, using propinol derivatives.

Keywords. Acridone alkaloids; Analogs of noracronycine; Chromene synthesis; Noracronycine; Mitsunobu-etherisation.

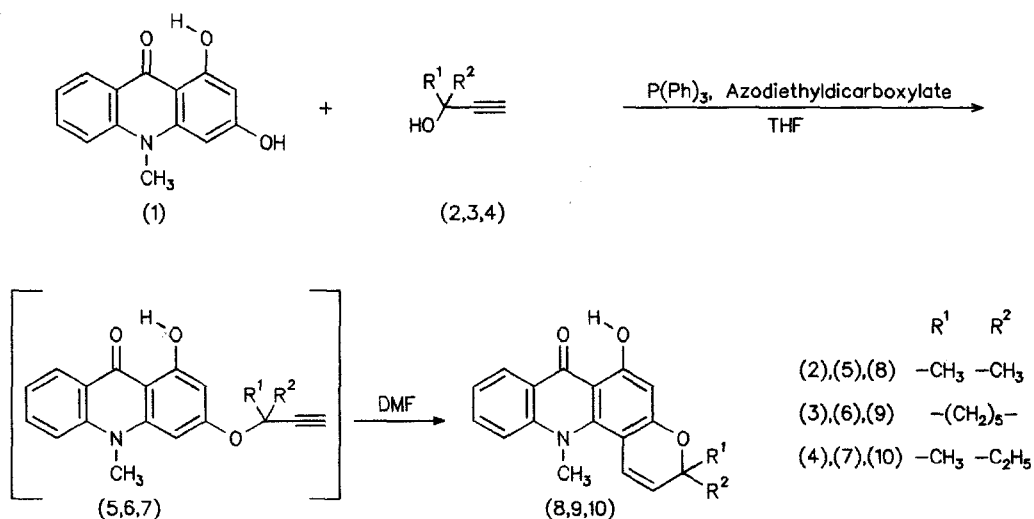
Acetylenchemie, 22. Mitt.: Synthese von Noracronycin und einiger Analoga mittels Mitsunobu-Reaktion (Kurze Mitt.)

Zusammenfassung. Der Naturstoff Noracronycin sowie einige seiner Analoga konnten durch Mitsunobu-Veretherung unter Verwendung von Propinol-Derivaten synthetisiert werden.

There is an increasing number of many natural products with a 2*H*-dimethylpyrane ring in their structure [3]. For the formation of this ring system, a strategy developed by Späth [4] (Späth's chromene synthesis) has been used. This concept has been varied in many ways without cognition of its origin (compare with [5, 6]). One of the disadvantages of these variations is the use of 1-dialkylated 1-chloropropynes for etherisation, since these substances are potentially unstable and with possible health hazards [7].

The direct etherisation of a phenolic group with an alcohol is possible if using the Mitsunobu-reaction [8]. Within the scope of our investigations into the structure-activity-relationship on noracronycine (**8**) [9, 10] this method was used for the first time in the synthesis of this natural product.

1,3-Dihydroxy-N-methyl-acridin-9(10H)-one (**1**) was easily prepared by a combination of synthetic methods [11, 12] and subsequent demethylation using 48% HBr. The reaction of **1** with the propinol derivatives **2–4** in the presence of triphenylphosphine and azodiethyldicarboxylate was carried out in *THF* [8]. The



resulting yields depend on the type of alcohol and were usually between 20 and 27%.

In case of the 1-ethynyl-1-cyclohexanol the linear isomer of the noracronycine derivative **9** could also be isolated, but its yield, compared with the amount of the angular form, was very low. This could be attributed to the type of solvent used [10].

It should be pointed out that some side products may be of great interest [10].

Experimental Part

In a 100 ml three-necked flask 482 mg (2 mmol) **1** was suspended in 25 ml abs. *THF* and 789 mg (3 mmol) triphenylphosphine was added. After subsequent addition of 3 mmol of a 1-dialkylated propynol derivative (**2**, **3** or **4**), the suspension was kept under nitrogen atmosphere. Then 502 mg (3 mmol) azodiethyl dicarboxylat, dissolved in 12 ml *THF*, was added dropwise (ca. 1 h). The reaction was stopped after 24 h and the solvent evaporated in vacuo. Column chromatography (CHCl₃) was used to separate the mixture of noracronycine derivative, the corresponding propargylether and other side products from excess reagents and starting materials. The final cyclisation of the propargylether to the corresponding noracronycine derivative was carried out by heating the mixture of products in *DMF* at 130°C for 5 h. The expected product was isolated by column chromatography (toluene).

Using 2-methyl-3-butyn-2-ol (**2**), noracronycine (**8**) (122 mg, 20%) was obtained, while with 1-ethynyl-1-cyclohexanol (**3**) spiro[6-hydroxy-12-methyl-3H-pyrano[2,3-c]acridin-7(12H)-on-3,1'-cyclohexane] (**9**) (212 mg, 27%) was isolated and 2-methyl-4-pentyn-2-ol (**4**) gave 157 mg (22%) 3-ethyl-6-hydroxy-3,12-dimethyl-3H-pyrano[2,3-c]acridin-9(10H)-one (**10**).

Noracronycine (**8**)

The spectroscopic data of **8** were identical with those given in [13].

Spiro[6-hydroxy-12-methyl-3H-pyrano[2,3-c]acridin-7(12H)-on-3,1'-cyclohexane] (**9**)

R_f : 0.35 (CHCl₃), 0.21 (toluene). M.p. 167°C. IR (KBr): $\nu = 3440 \text{ cm}^{-1}$ (br., 6-OH), 2930, 2860 (ν , C-H), 1640 (s, C=O), 1585, 1555, 1540 (m, arom. C=C), 1355 (m, C-H), 1275 (s, OH), 1135 (s, C-O), 745 (s, 4 vic. arom. H). UV (methanol): λ_{max} (log ϵ) = 418.0 nm (3.404), 286.0 (4.441), 256.5

(4.346), 228.0 (4.098). ¹H-NMR (CDCl₃): δ = 1.60 (m, 10 H, 2'-H, 3'-H, 4'-H, 5'-H, 6'-H), 3.90 (s, 3 H, N-CH₃), 5.53 (d, $J_{2-1} = 10.64$ Hz, 1 H, 2-H), 6.30 (s, 1 H, 5-H), 6.56 (d, $J_{1-2} = 10.64$ Hz, 1 H, 1-H), 7.30 (dd, $J_{9-8} = 8.01$ Hz, $J_{9-10} = 7.64$ Hz, 1 H, 9-H), 7.42 (d, $J_{11-10} = 8.65$ Hz, 1 H, H-11), 7.71 (dd, $J_{10-9} = 7.64$ Hz, $J_{10-11} = 8.65$ Hz, 1 H, 10-H), 8.37 (dd, $J_{8-9} = 8.01$ Hz, $J_{8-10} = 1.30$ Hz, 1 H, 8-H), 14.72 (s, 1 H, 6-OH). ¹³C-NMR (CDCl₃): δ = 21.7 (C-4'), 25.3 (C-3', C-5'), 35.2 (C-2', C-6'), 43.6 (N-CH₃), 76.4 (C-3), 97.9 (C-5), 101.6 (C-12 b), 107.3 (C-6 a), 116.1 (C-11), 121.7, 122.1, 122.4 (C-2, C-9, C-1), 126.3 (C-8), 133.9 (C-10), 144.4 (C-12 a), 144.9 (C-11 a), 161.5 (C-6), 165.3 (C-4 a), 181.2 (C-7). MS (70 eV): m/z (%) = 347 (42), 304 (100), 289 (21), 276 (10), 234 (8), 152 (8), 102 (4), 77 (9). C₂₂H₂₁NO₃: calc. 347.152144; found 347.1512648.

3-Ethyl-6-hydroxy-3,12-dimethyl-3 H-pyrano[2,3-c]acridin-7(12 H)-one (10)

*R*_f: 0.32 (CHCl₃), 0.20 (toluene). M.p. 183°C. IR (KBr): ν = 3440 cm⁻¹ (br., OH), 3000, 2940 (ν, C-H), 1635 (s, C=O), 1595, 1555, 1505 (m, arom. C=C), 1460 (m, C-H), 1330 (s, OH), 1150 (s, C-O), 745 (s, 4 vic. arom. H). UV (methanol): λ_{max} (log ε) = 405.0 nm (3.643), 270.5 (4.570), 223.5 (4.217). ¹H-NMR (CDCl₃): δ = 0.99 (t, $J_{14-13} = 7.5$ Hz, 3 H, 14-H), 1.47 (s, 3 H, 15-H), 1.84 (q, $J_{13-14} = 7.5$ Hz, 2 H, 13-H), 3.89 (s, 3 H, N-CH₃), 5.47 (d, $J_{2-1} = 9.7$ Hz, 1 H, 2-H), 6.25 (s, 1 H, 5-H), 6.57 (d, $J_{1-2} = 9.7$ Hz, 1 H, 1-H), 7.27 (dd, $J_{9-10} = 7.7$ Hz, $J_{9-8} = 8.1$ Hz, 1 H, 9-H), 7.40 (dd, $J_{11-10} = 8.6$ Hz, 1 H, 11-H), 7.69 (ddd, $J_{10-11} = 8.6$ Hz, $J_{10-9} = 7.7$ Hz, $J_{10-8} = 1.6$ Hz, 1 H, 10-H), 8.34 (dd, $J_{8-9} = 8.1$ Hz, $J_{8-10} = 1.6$ Hz, 1 H, 8-H), 14.72 (s, 1 H, 6-OH). ¹³C-NMR (CDCl₃): δ = 8.2 (C-14), 24.6 (C-15), 32.8 (C-13), 43.7 (N-CH₃), 79.0 (C-3), 97.8 (C-5), 101.0 (C-12 b), 106.9 (C-6 a), 116.3 (C-11), 121.8, 121.9, 122.0 (C-2, C-9, C-1), 126.2 (C-8), 133.9 (C-10), 144.3 (C-12 a), 144.9 (C-11 a), 161.8 (C-6), 165.2 (C-4 a), 181.1 (C-7). MS: m/z (%) = 321 (27) [*M*⁺], 306 (19), 292 (100), 278 (32), 264 (14), 250 (10), 241 (53), 212 (31), 184 (14), 140 (9), 77 (20). C₂₀H₁₉NO₃: calc. 321.136494; found 321.137368.

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Received October 16, 1991. Accepted November 5, 1991